

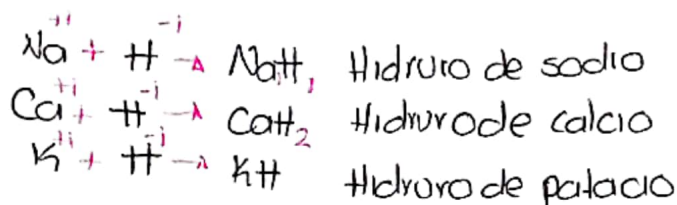


INSTITUCIÓN EDUCATIVA MIGUEL DE CERVANTES SAAVEDRA

Taller de refuerzo de nomenclatura inorgánica.

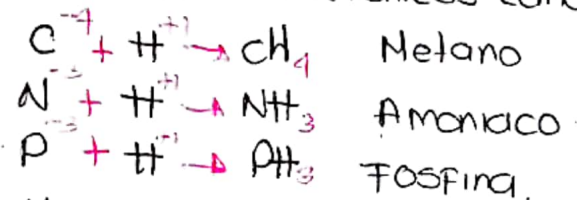
De ejemplos de
Hidruros Metálicos;
Metálicos

IA	IIA	IIIA	IVA	VA	VIA	VIIA
1	2	3	4	5	6	7

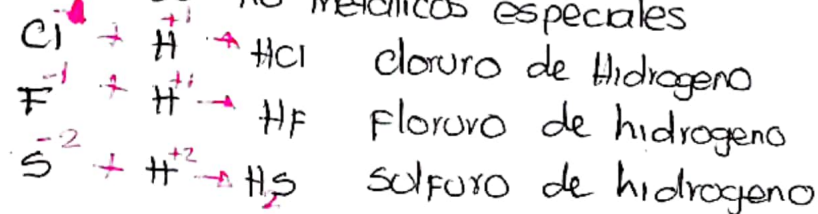


Hidruros no metálicos se agrupan de dos común y especial

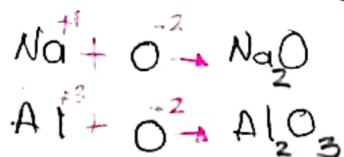
Hidruros no metálicos común



Hidruros no metálicos especiales

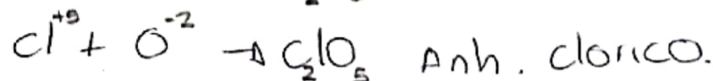
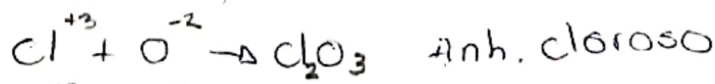


Oxido Metálico

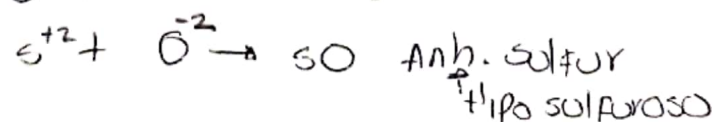
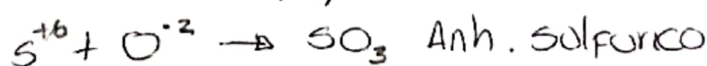


Oxido no Metálico

val (Cl): 1, 3, 5, 7



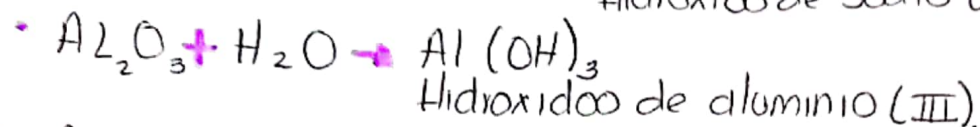
val (S): 2, 4, 6



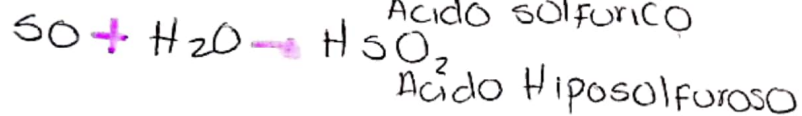
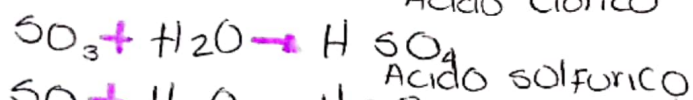
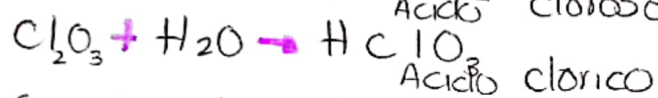
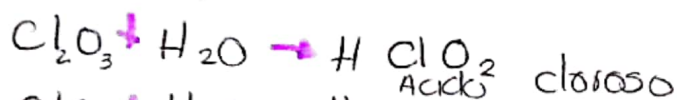


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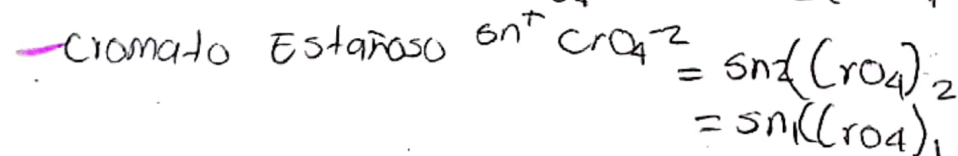
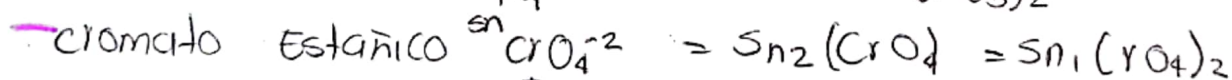
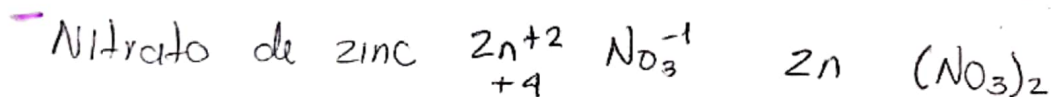
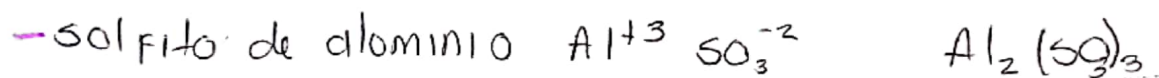
• Hidroxidos.



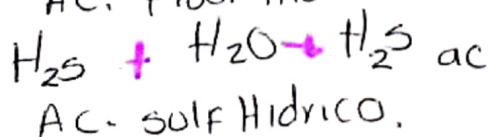
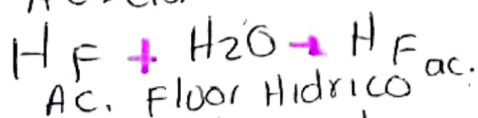
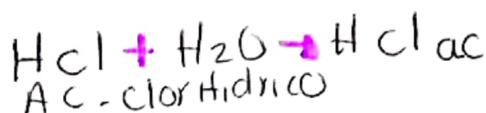
• Ac. OXACIDO.



• Sal oxisal.



• Acido Hidroacido

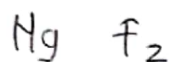
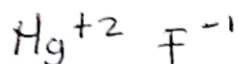




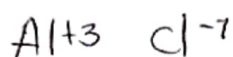
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• Sales Halaideas.

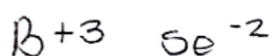
1.- Fluoruro de Magnesio.



2. cloruro de Aluminio



3.- seleniuro de boro



4. Fosforo de cobalto (III)

