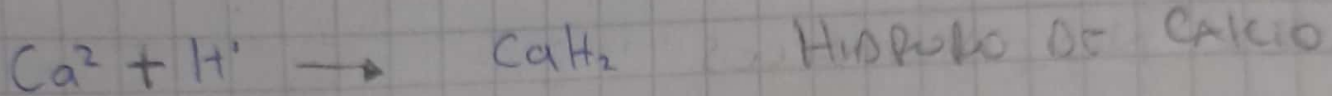
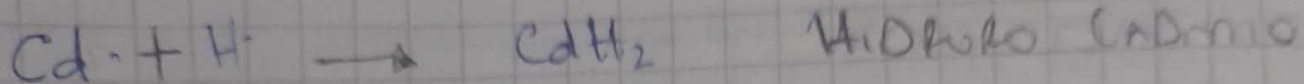
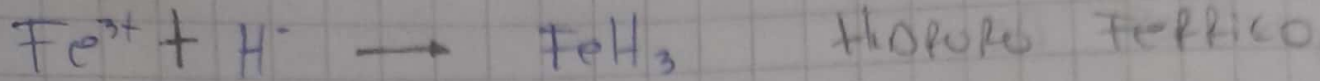
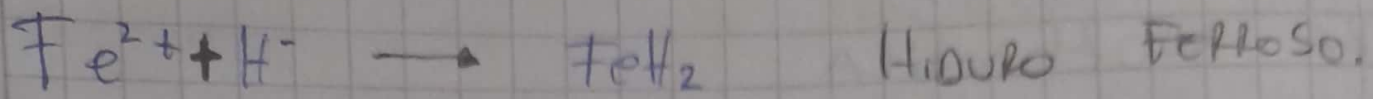


# Taller DE REPETICIÓN DE NOMENCLATURA INORGÁNICA

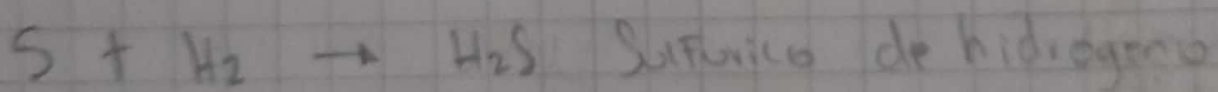
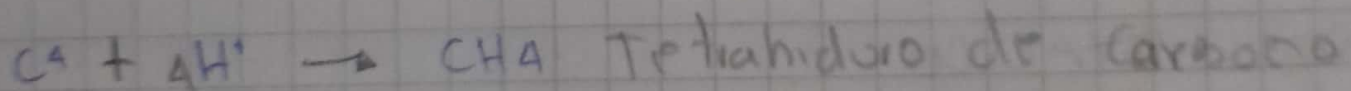
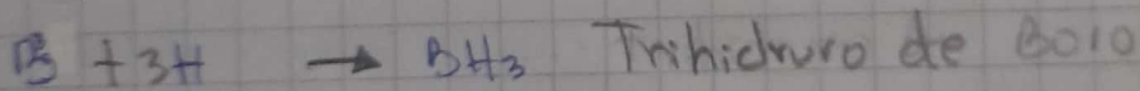
DE 5 EJEMPLOS DE

- HIDRÓXIDOS METÁLICOS
- HIDRÓXIDOS NO METÁLICOS
- HIDRÓXIDO NO METÁLICO COMÚN
- HIDRÓXIDO NO METÁLICO ESPECIAL
- ÓXIDOS METÁLICOS
- ÓXIDOS NO METÁLICOS
- HIDRÓXIDO
- ÁCIDO ÓXÁCIDO
- SAL ÓXISAL
- ÁCIDO HIDRÁCIDO
- SAL ALÓIDEA

## "Solución"



## HIDRuros NO METÁLICOS



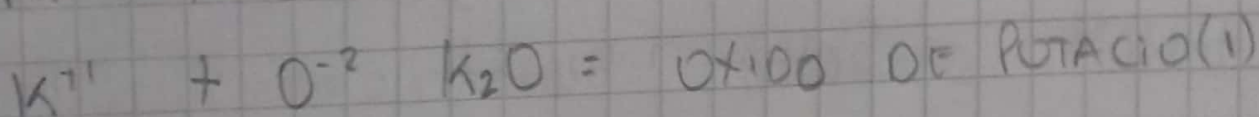
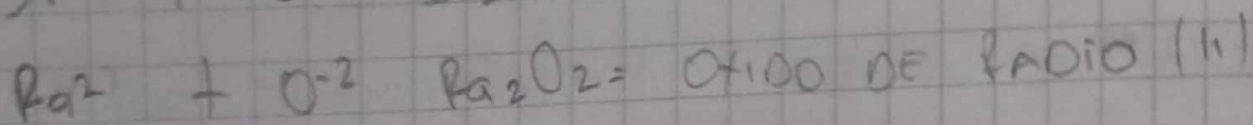
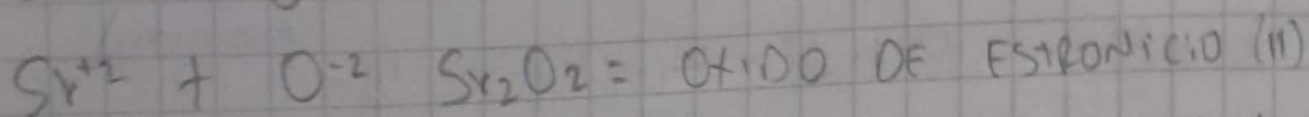
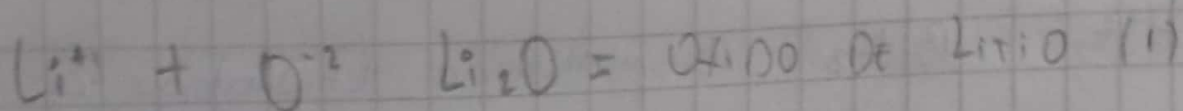
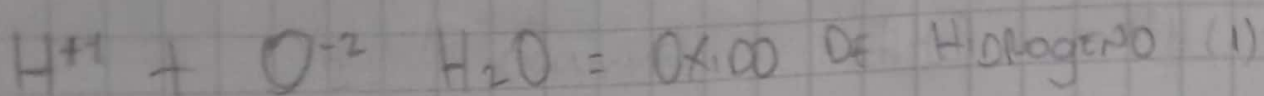
## HIDROGENO NO METALICOS COMUN

|                    |         |                                    |
|--------------------|---------|------------------------------------|
| $Ge^{-4} + H^{+1}$ | $GeH_4$ | Germano                            |
| $Bi^{-3} + H^{+1}$ | $BiH_3$ | Bismotirina (hidrogeno de bismuto) |
| $Si^{-4} + H^{+1}$ | $SiH_4$ | Hidrogeno de silicio               |
| $B^{-3} + H^{+1}$  | $BH_3$  | Hidrogeno de boro                  |
| $Sb^{-3} + H^{+1}$ | $SbH_3$ | Hidrogeno de antimonio             |

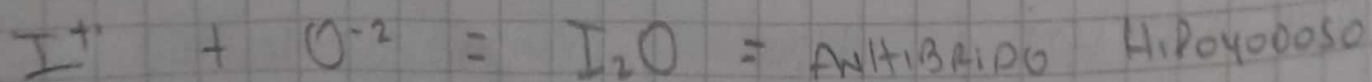
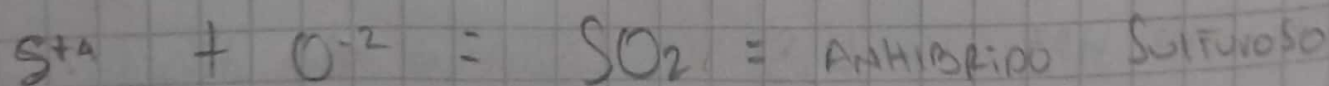
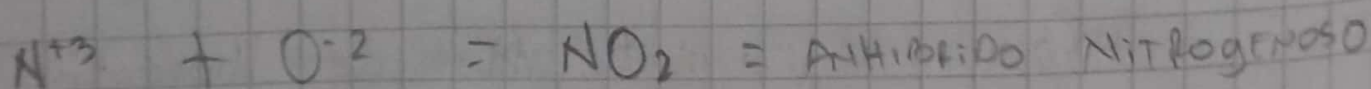
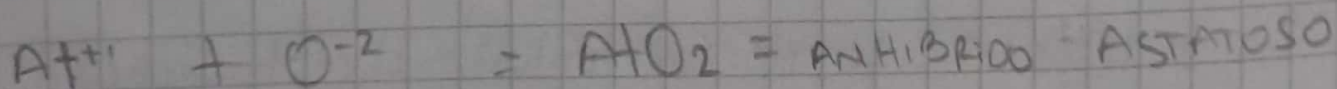
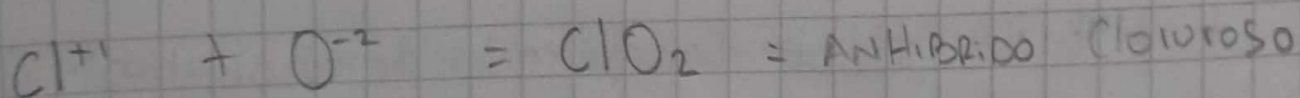
## HIDROGENO NO METALICO ESPECIAL

|                    |         |                      |
|--------------------|---------|----------------------|
| $Se^{-2} + H^{+1}$ | $H_2Se$ | Selenio de hidrogeno |
| $Te^{-2} + H^{+1}$ | $H_2Te$ | telurio de hidrogeno |
| $I^{-1} + H^{+1}$  | $HI$    | yoduro de hidrogeno  |
| $S^{-2} + H^{+1}$  | $H_2S$  | Sulfuro de hidrogeno |
| $Br^{-1} + H^{+1}$ | $HBr$   | Bromuro de hidrogeno |

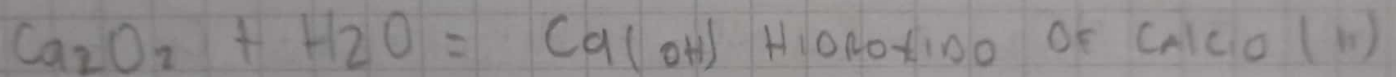
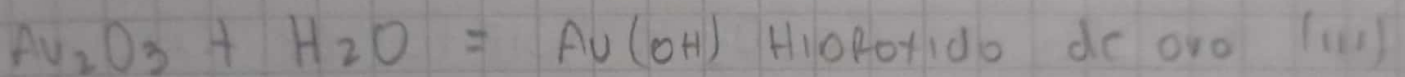
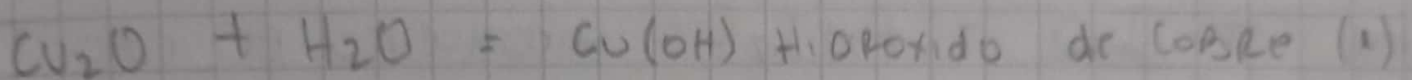
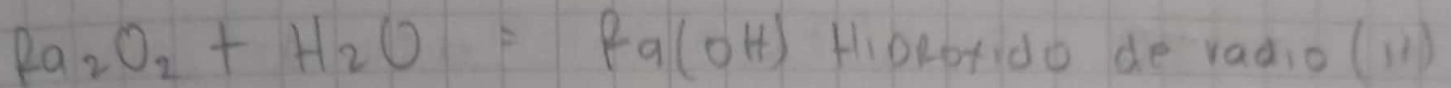
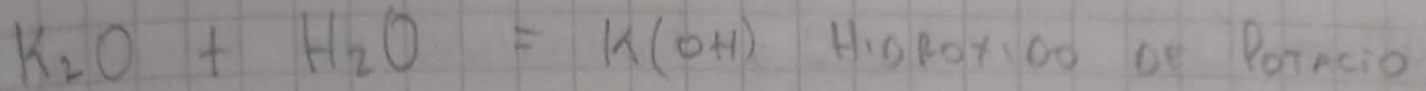
## Oxidos metálicos



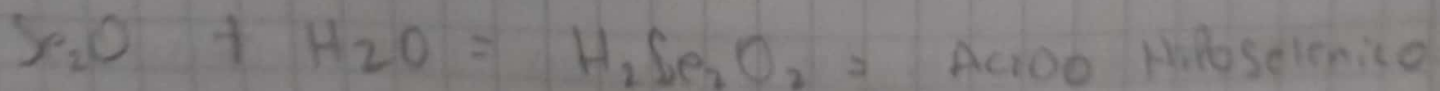
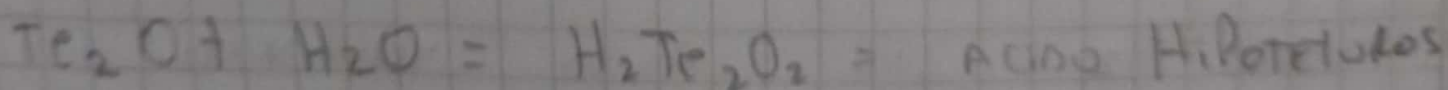
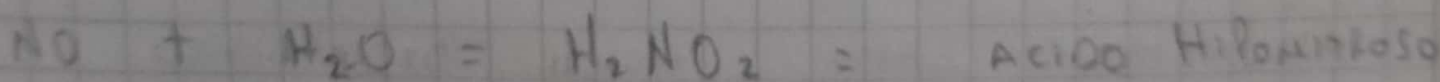
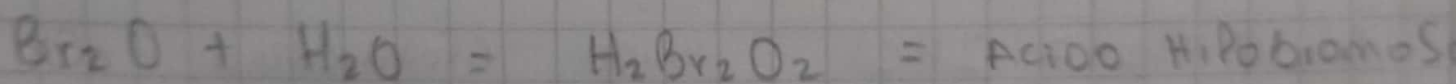
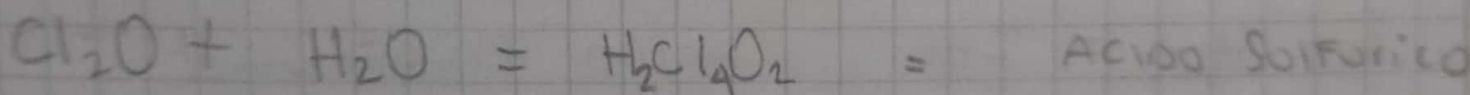
## Oxidos NO METÁLICOS (ANHIDRIDOS)



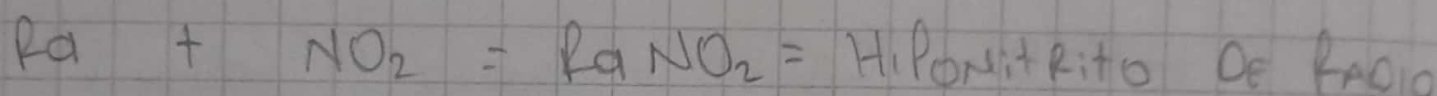
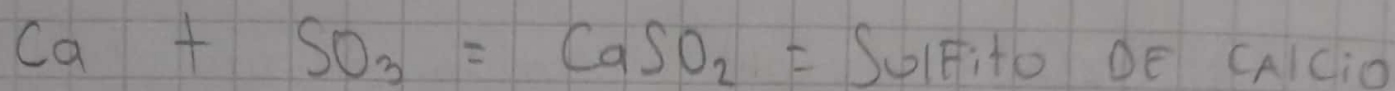
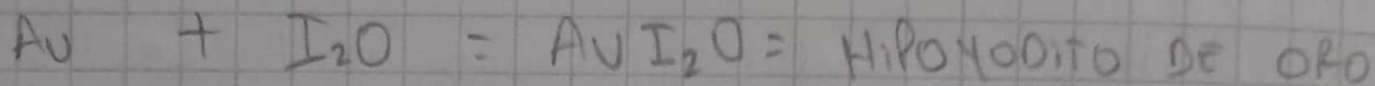
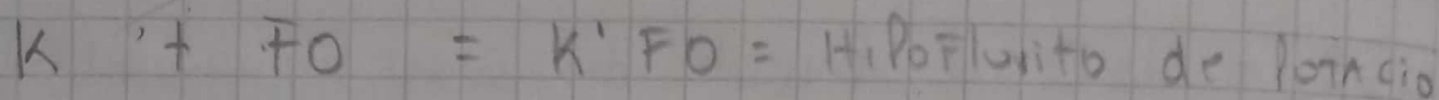
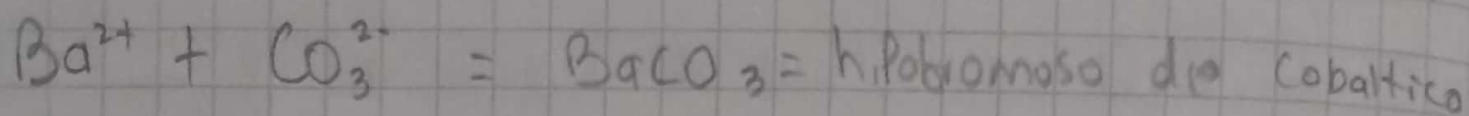
## "Hidróxidos"



## ÁCIDO OTÁCIDO.



## "Sal Oxisal"



## Acido Hidroacido

