

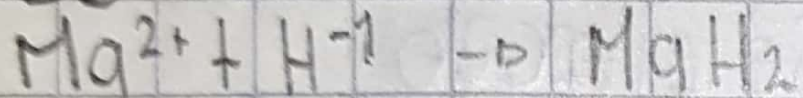
teniendo en cuenta el video visto en clase.

• De 5 ejemplos.

### \* Hidruros Metálicos.



Hidruro de Berilio



Hidruro de Magnesio



Hidruro de Aluminio

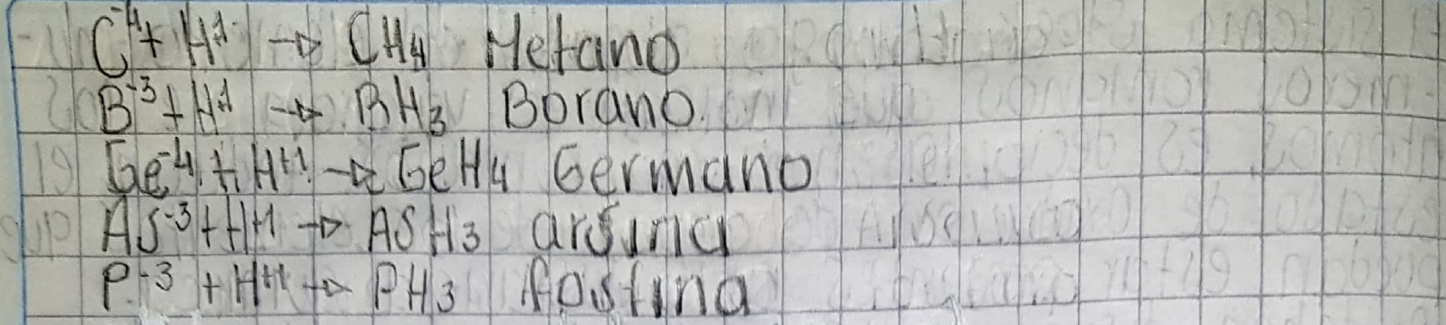


Hidruro de Bario

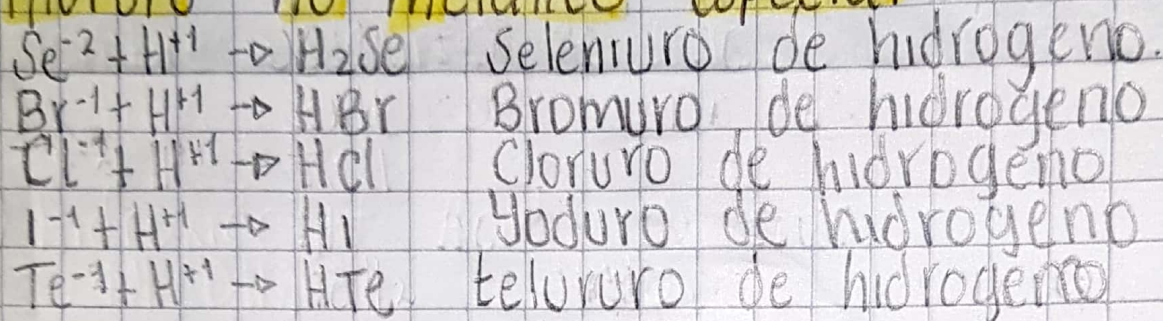


Hidruro de Litio.

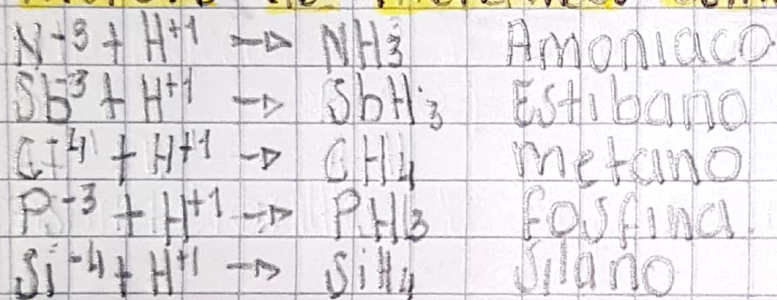
### \* Hidruros no Metálicos:



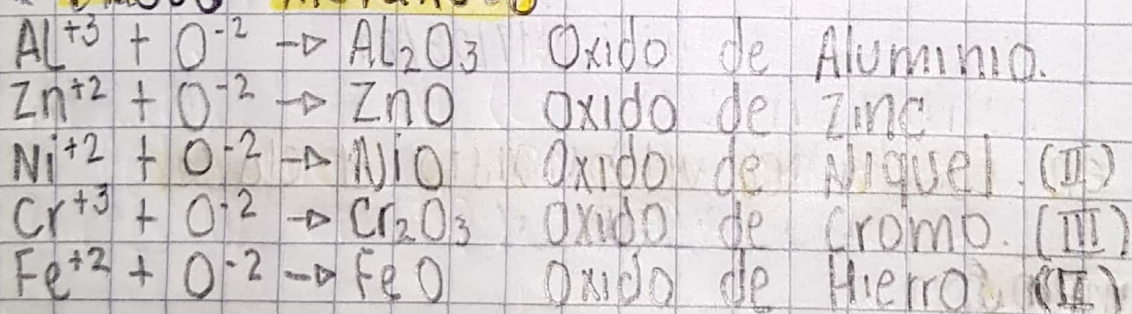
### Hidruro no metálico especial



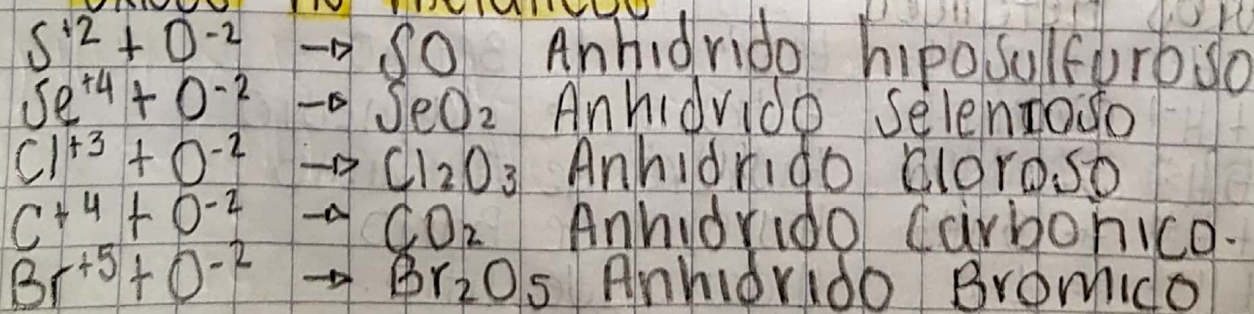
### \* Hidruro no metálicos Común



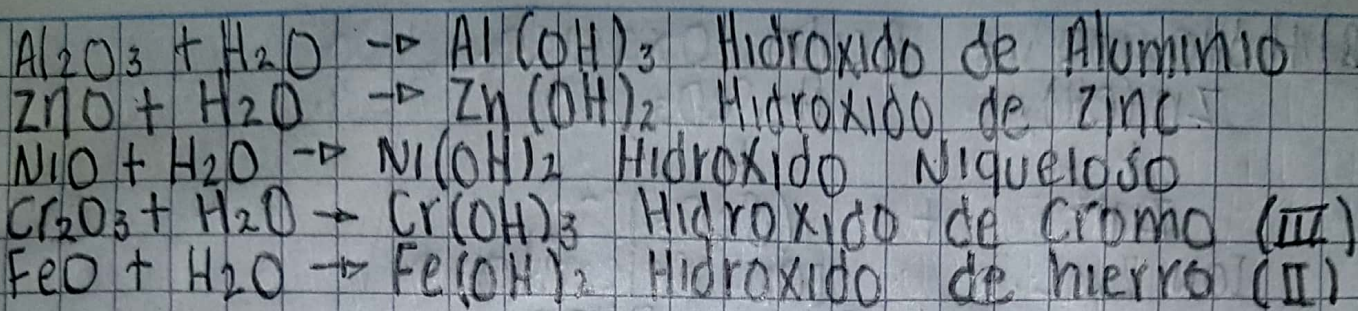
### \* Óxidos metálicos



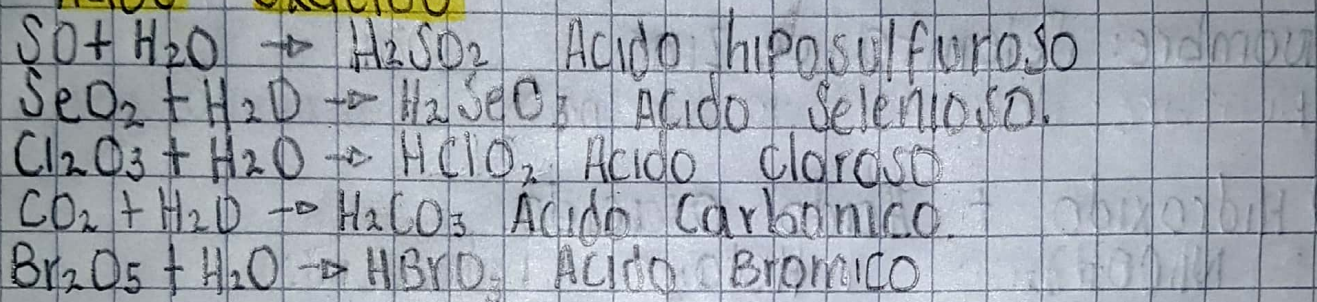
### \* Óxidos no metálicos



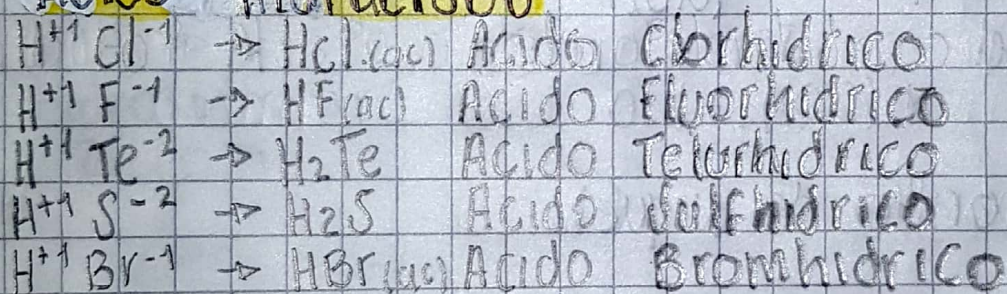
### \* Hidróxido



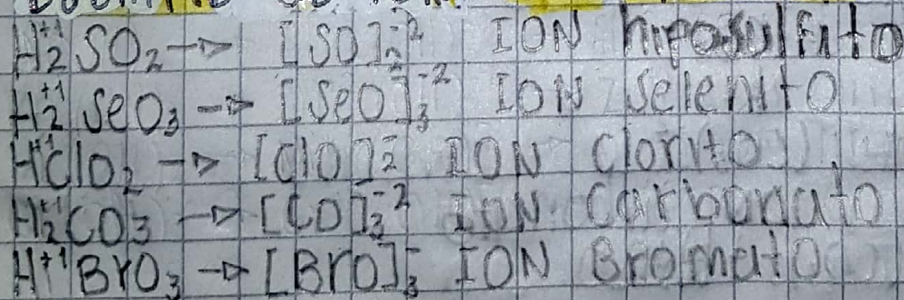
### \* Acido Oxacido



### \* Acido Hidracidos

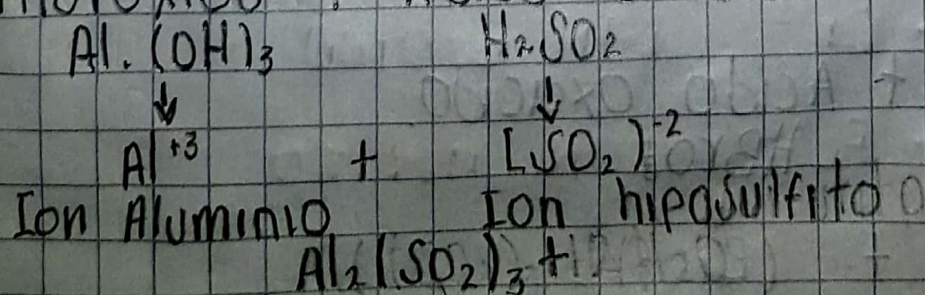


### \* Ejemplo de ION NO metalico.



### \* Sal Oxisal

1) Hidroxido + Acido Oxacido.



Nombre: hiposulfito de Aluminio.