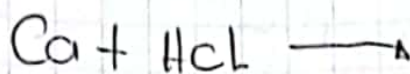


Química ciclo VI

Solución

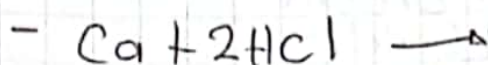


Reactantes

$$\text{Ca} = 1$$

$$\text{H} = 1$$

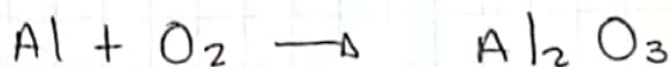
$$\text{Cl} = 1$$



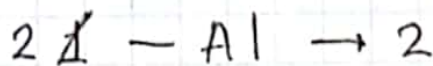
$$\text{Ca} = 1$$

$$\text{H} = 2$$

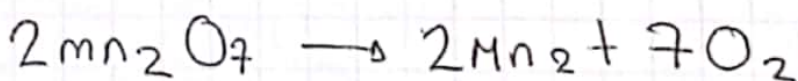
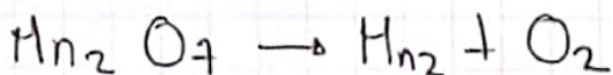
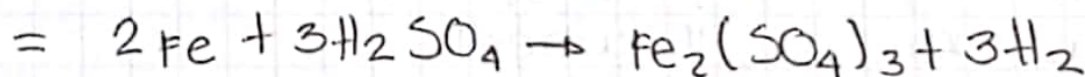
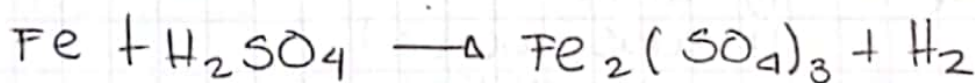
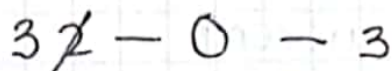
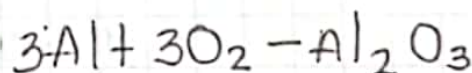
$$\text{Cl} = 2$$

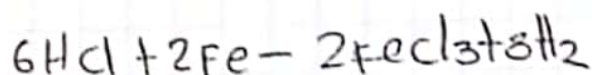
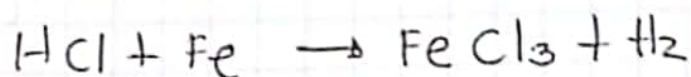
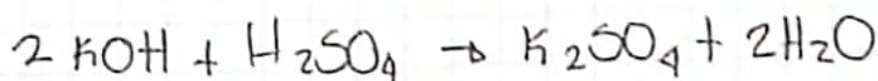
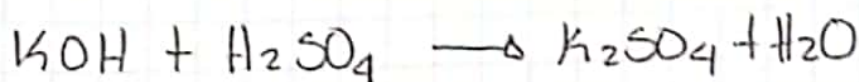
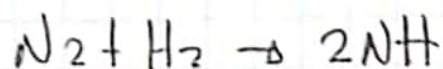
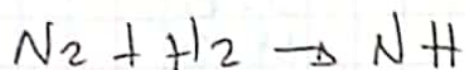
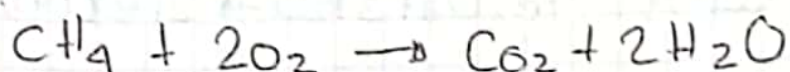
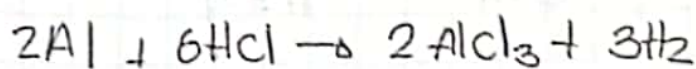
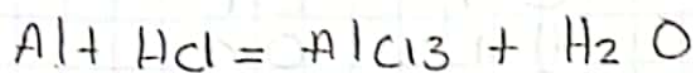
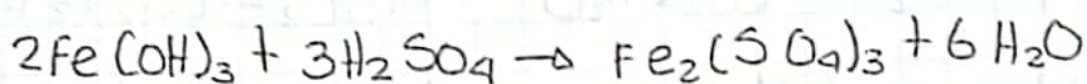
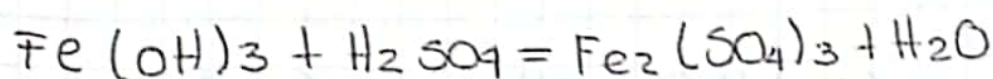
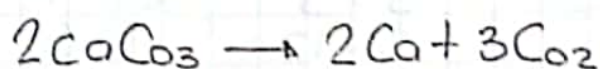
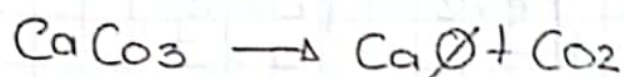
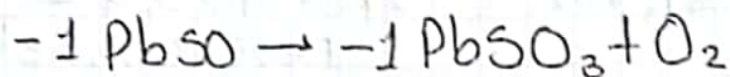
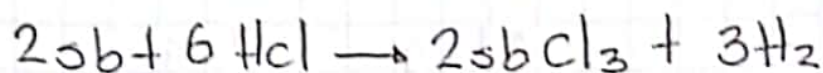
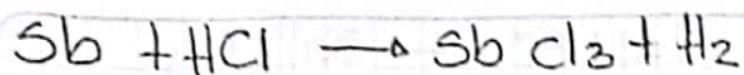


R



P





$$\text{H} = 1$$

$$\text{Cl} = 1$$

$$\text{Fe} = 1$$

$$\text{Fe} = 1$$

$$\text{Cl} = 3$$

$$\text{H} = 2$$

$$\text{H} = 6$$

$$\text{Cl} = 6$$

$$\text{Fe} = 2$$

$$\text{Fe} = 2$$

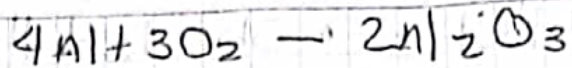
$$\text{Cl} = 6$$

$$\text{H} = 6$$



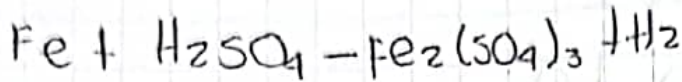
$$Al = 1 \\ O = 2$$

$$Al = 2 \\ O = 3$$



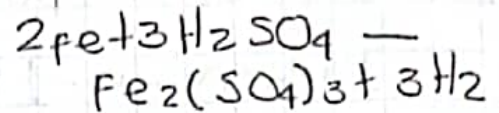
$$Al = 4 \\ O = 6$$

$$Al = 4 \\ O = 6$$



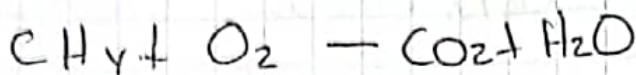
$$Fe = 1 \\ H = 2 \\ S = 1 \\ O = 4$$

$$Fe = 2 \\ S = 3 \\ O = 4 \\ H = 2$$



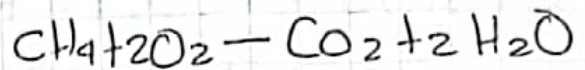
$$Fe = 2 \\ H = 6 \\ S = 3 \\ O = 12$$

$$Fe = 2 \\ S = 3 \\ O = 12 \\ H = 6$$



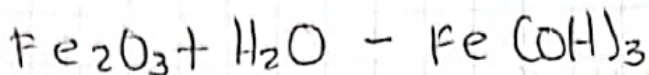
$$C = 1 \\ H = 4 \\ O = 2$$

$$C = 1 \\ O = 3 \\ H = 2$$



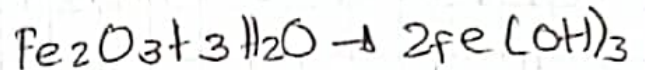
$$C = 1 \\ H = 4 \\ O = 4$$

$$C = 1 \\ O = 4 \\ H = 4$$



$$Fe = 2 \\ O = 4 \\ H = 2$$

$$Fe = 1 \\ O = 3 \\ H = 3$$



$$Fe = 2 \\ O = 6 \\ H = 6$$

$$Fe = 2 \\ O = 6 \\ H = 6$$